

Exhibit B

# TRANSPORT PHENOMENA

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## Preface

This book is intended to present the subjects of transport (heat conduction, port (diffusion). In this phenomena are occurring & said about the molecular & continuum approach is of & dents, although it should be for complete mastery of the

Because of the current de emphasis on understanding use of empiricism, we feel tl kind. Obviously the subje across traditional departme subject of transport phenom mechanics, and electromag ences." Knowledge of the transport has certainly bec gineering analysis. In add interest to some who are & meteorology, and biology.

L CONDUCTIVITY OF SOME  
RIC PRESSURE<sup>a</sup>

ure )	Thermal Conductivity $k$ (cal sec <sup>-1</sup> cm <sup>-1</sup> (° K) <sup>-1</sup> )
	0.000378
	0.000363
	0.000247
	0.000328
	0.000400
	0.000703
	0.00143
	0.00156
	0.00160

ysics, Thirty-ninth Edition, Chemical  
), pp. 2257-2259. Corrected data for  
thers.

AL CONDUCTIVITIES OF  
PHERIC PRESSURE<sup>a</sup>

e	Thermal Conductivity $k$ (cal sec <sup>-1</sup> cm <sup>-1</sup> (° K) <sup>-1</sup> )
	0.247
	0.290
	0.106
	0.119
	0.039
	0.037
	0.036
	0.0196
	0.0261
	0.0303
	0.1073
	0.0956
	0.0846
	0.2055
	0.1809
	0.1596
	0.0617
	0.0648
	0.0675

Vol. 2, Atomic Energy Commission,  
fice, Washington, D.C. (May, 1955),

TABLE 8.1-4  
EXPERIMENTAL VALUES OF THERMAL CONDUCTIVITIES OF  
SOME SOLIDS<sup>a</sup>

Substance	Temperature $T$ (° C)	Thermal Conductivity $k$ (cal sec <sup>-1</sup> cm <sup>-1</sup> (° K) <sup>-1</sup> )
Aluminum	100	0.492
	300	0.64
	600	1.01
Cadmium	0	0.220
	100	0.216
Copper	18	0.918
	100	0.908
Steel	18	0.112
	100	0.107
Tin	0	0.1528
	100	0.143
Brick (common red)	—	0.0015
Concrete (stone)	—	0.0022
Earth's crust (av.)	—	0.004
Glass (soda)	200	0.0017
Graphite	—	0.012
Sand (dry)	—	0.00093
Wood (fir)		
parallel to axis	—	0.00030
perpendicular to axis	—	0.00009

<sup>a</sup> Data taken from the *Reactor Handbook*, Vol. 2, Atomic Energy Commission, AECD-3646, U.S. Government Printing Office, Washington, D.C. (May, 1955), pp. 1766 *et seq.*

Substitution in Eq. 8.1-1 then gives

$$k = \frac{QY}{A \Delta T} = \frac{0.717 \times 0.640 \text{ (cal sec}^{-1}\text{)(cm)}}{929 \times 2.00 \text{ (cm}^2\text{)(}^\circ\text{K)}} \\ = 2.47 \times 10^{-4} \text{ cal sec}^{-1} \text{ cm}^{-1} \text{ (}^\circ\text{K)}^{-1}$$

For  $\Delta T$  as small as this, it is usually reasonable to assume that the value of  $k$  applies at the average temperature  $(T_1 + T_0)/2$ , which in this case is 25° C. See Problems 9.F and 9.J for methods of allowing for variation of  $k$  with  $T$ .

§8.2 TEMPERATURE AND PRESSURE DEPENDENCE OF THERMAL  
CONDUCTIVITY IN GASES AND LIQUIDS

The scarcity of reliable thermal conductivity data for fluids frequently makes it necessary to estimate  $k$  from other data on the given substance. We present here two correlations to aid in such estimation and to illustrate how

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